



Metrology for Climate Relevant VOCs

Dynamic reference gas mixture preparation and uncertainty: permeation and diffusion methods

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POLITECNICO
DI TORINO

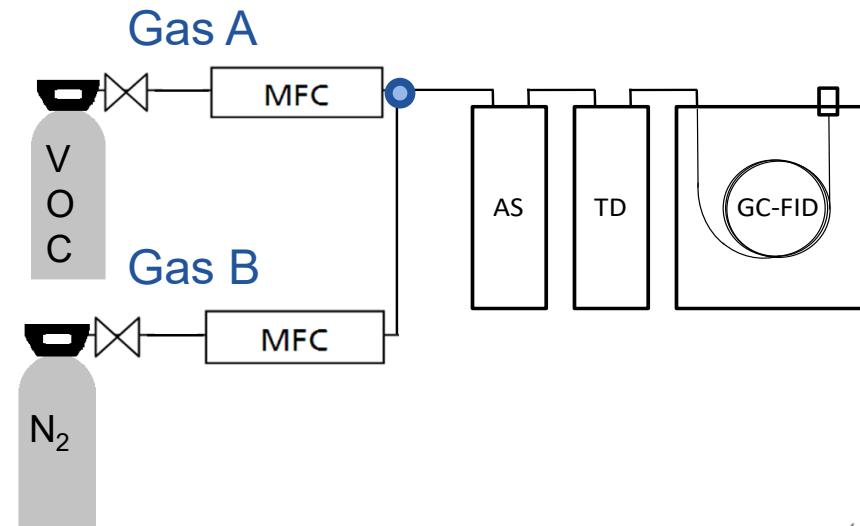


The EMPIR initiative is co-funded by the European Union's Horizon 2020 research and innovation programme and the EMPIR Participating States

ISO 6145 – dynamic methods

Preparation of calibration gas mixtures using dynamic methods, which rely in flow rates (gas A introduced at a known constant volume or mass flow rate into a known constant flow rate of gas B).

Dynamic dilution of a high fraction reference
gas mixture with mass-flow controllers (MFCs)



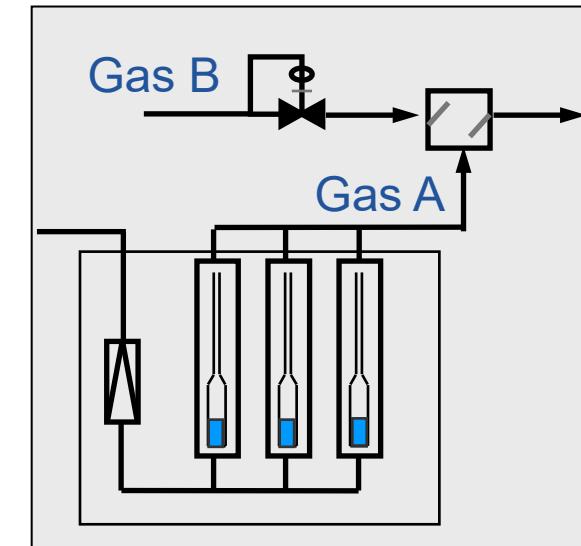
(Source: courtesy of VSL)

ISO 6145 – dynamic methods

Main dynamic methods

- Piston pumps
- Continuous injection
- Capillary
- Critical orifices
- Thermal mass-flow controllers
- **Diffusion**
- Saturation
- **Permeation**
- Electrochemical generation

Diffusion



(Source: courtesy of VSL)

Advantages and disadvantages

STATIC METHODS

VS.

DYNAMIC METHODS

- Wide range of concentrations.
- Un-tunability
- Suboptimal for unstable compounds
- Size and cost of cylinders
- Safety issues
- High minimal concentration
- Portability
- Limited long term stability

Advantages and disadvantages

STATIC METHODS VS.

DYNAMIC METHODS

- Wide range of concentrations
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- Suboptimal for unstable compounds
- Size and cost of cylinders
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- High minimal concentration
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- Wide range of concentrations
- Tunability
- Unstable compounds
- Long life low cost
- Safety issues
- Lower minimal concentration
- Low portability
- Long term stable

Permeation method

Permeation dynamic method: MSB



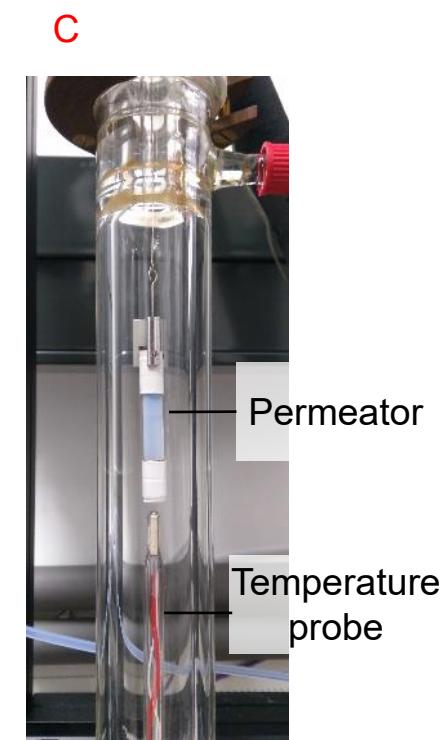
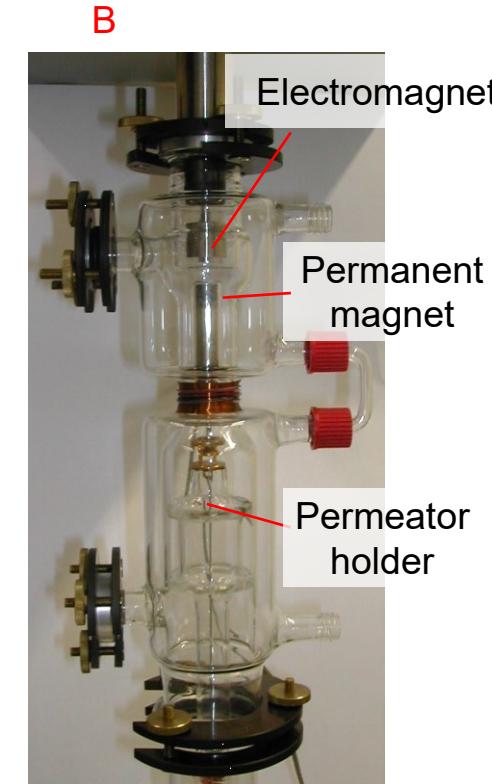
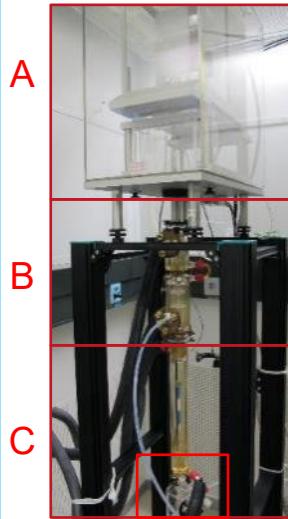
- TA Instruments (former Rubotherm)
- Glass/metal (SilcoNert 2000 coated SS)

Magnetic suspension balance (MSB):
glass (left), metal (right)

<https://www.metclimvoc.eu/training.html> (March 2021)

Permeation dynamic method: MSB

MSB elements

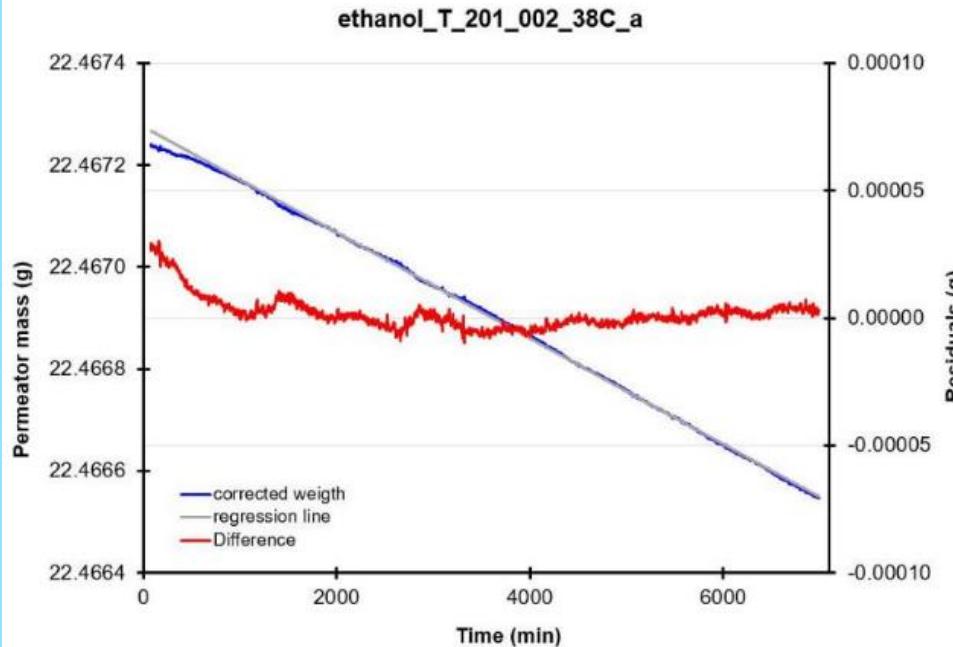


Permeation dynamic method: generation

Primary reference gas mixtures using a MSB (magnetic suspension balance)

Step 1: Calibration of the permeation unit

Controlled conditions (T, P, flow)

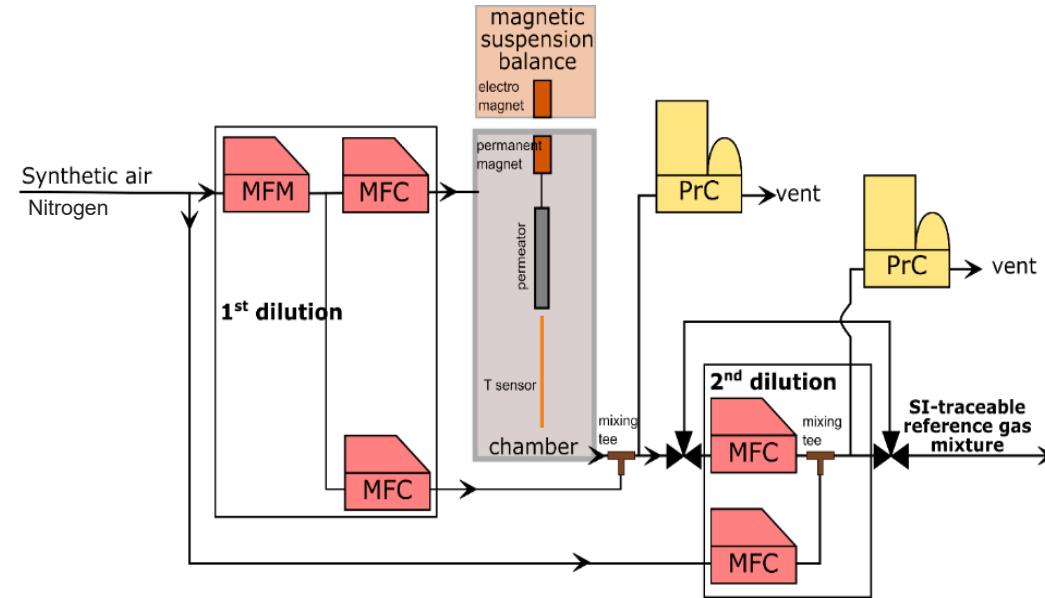


Permeation rate:

$$q_m = \frac{\Delta m}{\Delta t} = \frac{m_2 - m_1}{t_2 - t_1}$$

Permeation dynamic method: generation

Step 2: Dilution



$$1^{\text{st}} \text{ dilution: } X_{\text{RGM}} = q_m \cdot \text{purity} \cdot \left(\frac{V_{\text{mgas}}}{M_{\text{mcomp}}} \right) \cdot f_{\text{dil1}} + X_{\text{res}} \quad (> 30-50 \text{ nmol/mol})$$

$$2^{\text{nd}} \text{ dilution: } X_{\text{RGM}} = \left(q_m \cdot \text{purity} \cdot \left(\frac{V_{\text{mgas}}}{M_{\text{mcomp}}} \right) \cdot f_{\text{dil1}} + X_{\text{res}} \right) \cdot f_{\text{dil2}} + X_{\text{res}} \quad (< 30 \text{ nmol/mol})$$

Permeation dynamic method: uncertainty

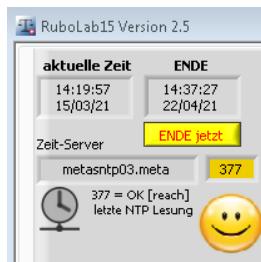
Main contributors to the uncertainty of the RGM amount fraction generated

$$X_{\text{RGM}} = q_m \cdot \text{purity} \cdot \left(\frac{V_{m_{\text{gas}}}}{M_{m_{\text{comp}}}} \right) \cdot f_{\text{dil1}} + X_{\text{res}}$$

Permeation rate (step 1)

$$q_m = \frac{\Delta m}{\Delta t}$$

- Balance sensitivity: < 1.5 % ($u(S) = 3 \cdot 10^{-5} \text{ g}$)
- Buoyancy variations: < 1 % ($u(b) < 3 \cdot 10^{-7} \text{ g}$)
- Chamber temperature variations: negligible ($u(T) < 0.00026 \text{ }^{\circ}\text{C}$)
- Chamber pressure variations: negligible ($u(P) < 0.002 \text{ hPa}$)
- Noise of the system > 75 % ($u(SC) < 0.004 \text{ \%}$)



- PC time synchronized through a Network-Time-Protocol (NTP) server with the Swiss official time given by **atomic clocks** at METAS photonic, time and frequency lab (**negligible**; $u(t) < 1 \cdot 10^{-18} \text{ s}$)

- Others: leaks, wall reactions...

Permeation dynamic method: uncertainty

Main contributors to the uncertainty of the RGM amount fraction generated

$$X_{\text{compound}} = q_m \cdot \text{purity} \cdot \left(\frac{V_{m_{\text{gas}}}}{M_{m_{\text{comp}}}} \right) \cdot f_{\text{dil1}} + X_{\text{res}}$$

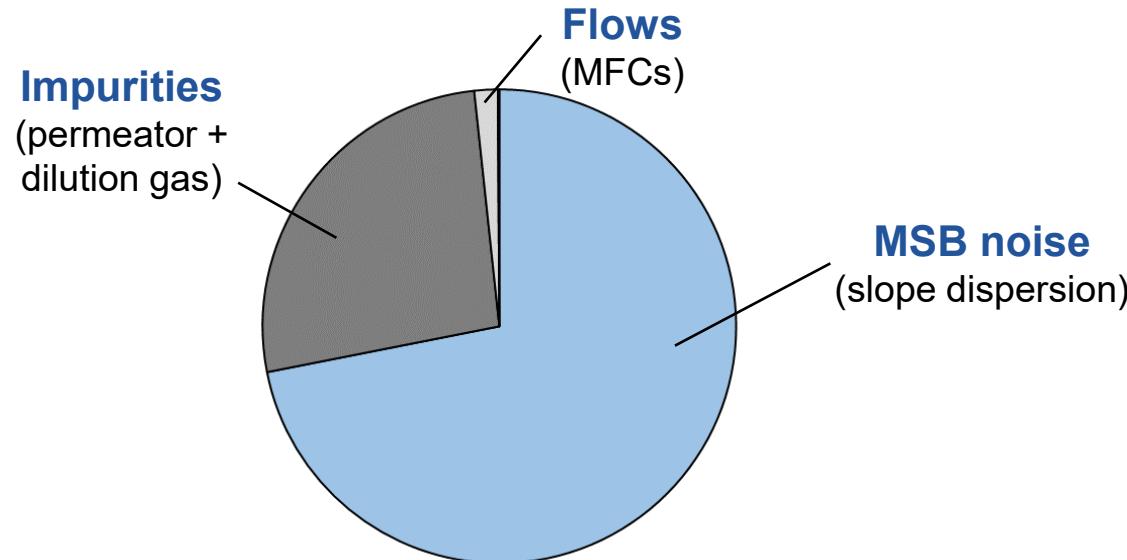
Step 2

- Impurities: variable
- Molar mass, molar volume: negligible (at 10^{-6} nmol/mol level)
- Dilution flow: 4-10 % ($u(q_v) < 0.2 \%$)
- Residuals in dilution gas: variable
- Others: leaks, wall reactions...

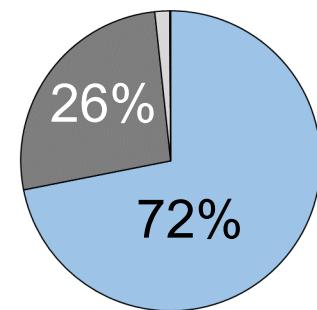
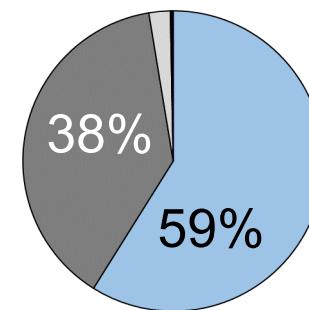
Permeation dynamic method: uncertainty

$$U(X_{RGM}) = k \cdot u_c(X_{RGM})$$

$$\frac{u_c(X_{RGM})}{X_{RGM}} = \sqrt{\left(\frac{u(q_m)}{q_m}\right)^2 + \left(\frac{u(q_v)}{q_v}\right)^2 + \dots +}$$



Permeator	Dilution steps	Amount fraction (nmol/mol)	Uncertainty ($k = 2$)
Perm1	2	31.3	1.9%
Perm2	1	345.9	1.6%



Perm1

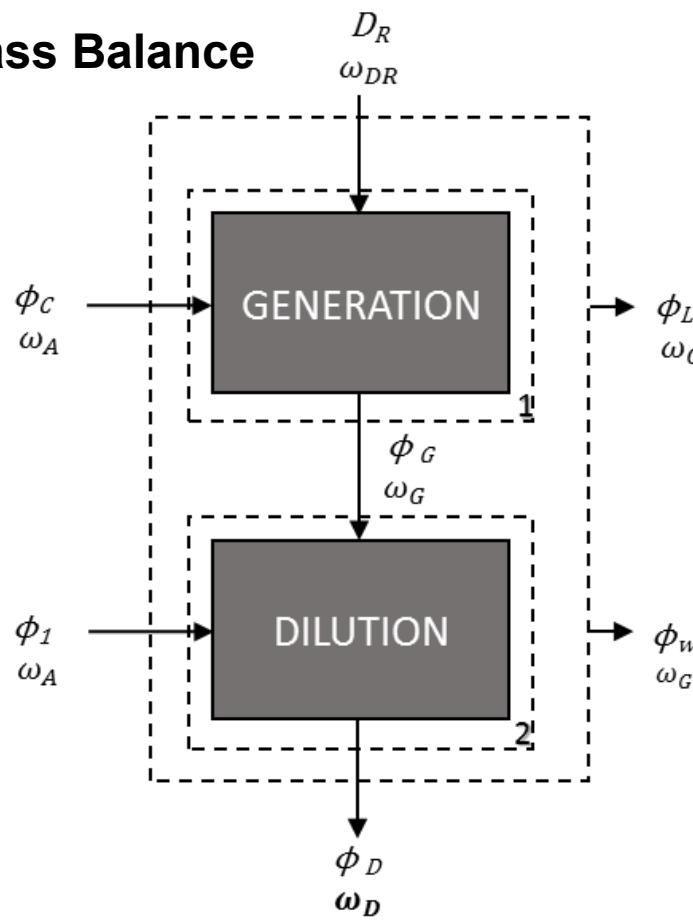
Perm2



Diffusion method

Diffusion dynamic method

Mass Balance

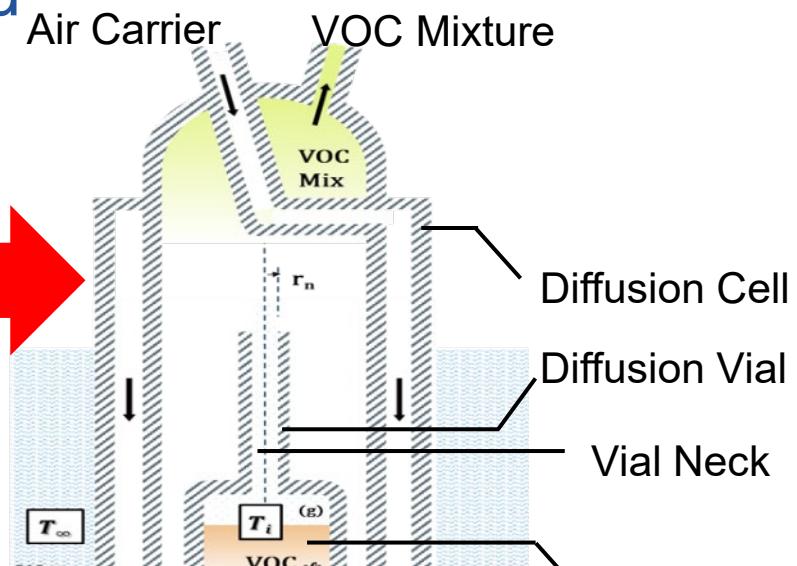
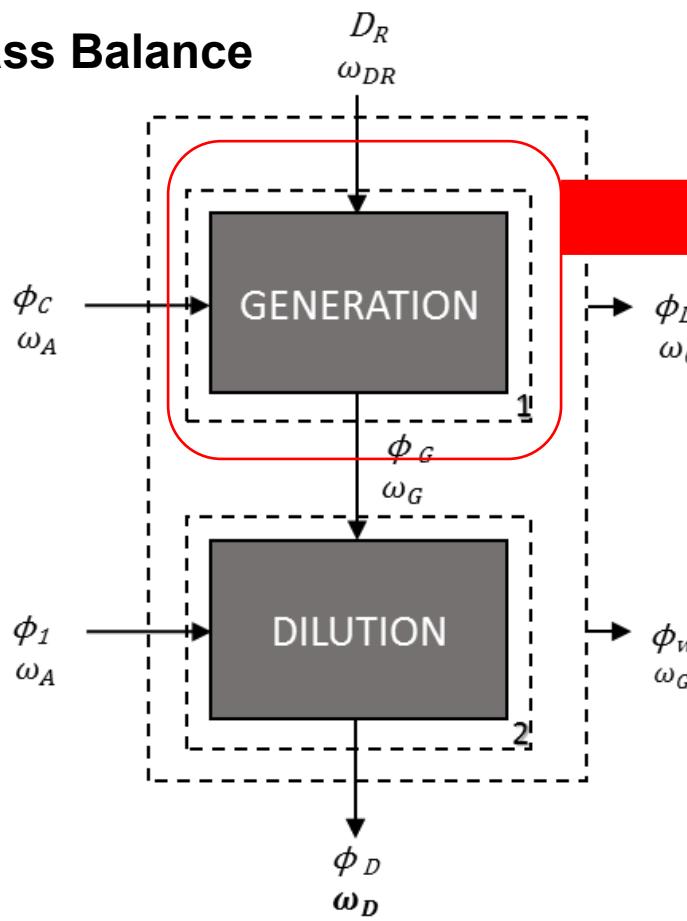


$P_v = 1-70 \text{ kPa} @ \text{generation T } (26^\circ\text{C})$

- ✓ Fick's law of diffusion
- ✓ Perfect gas assumption
- ✓ Convective transport
- ✓ Turbulence for enhanced mixing

Diffusion dynamic method

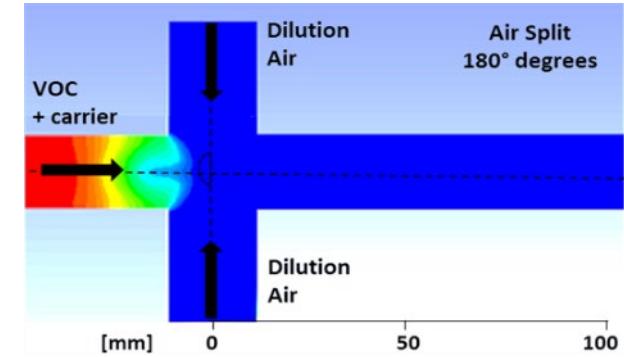
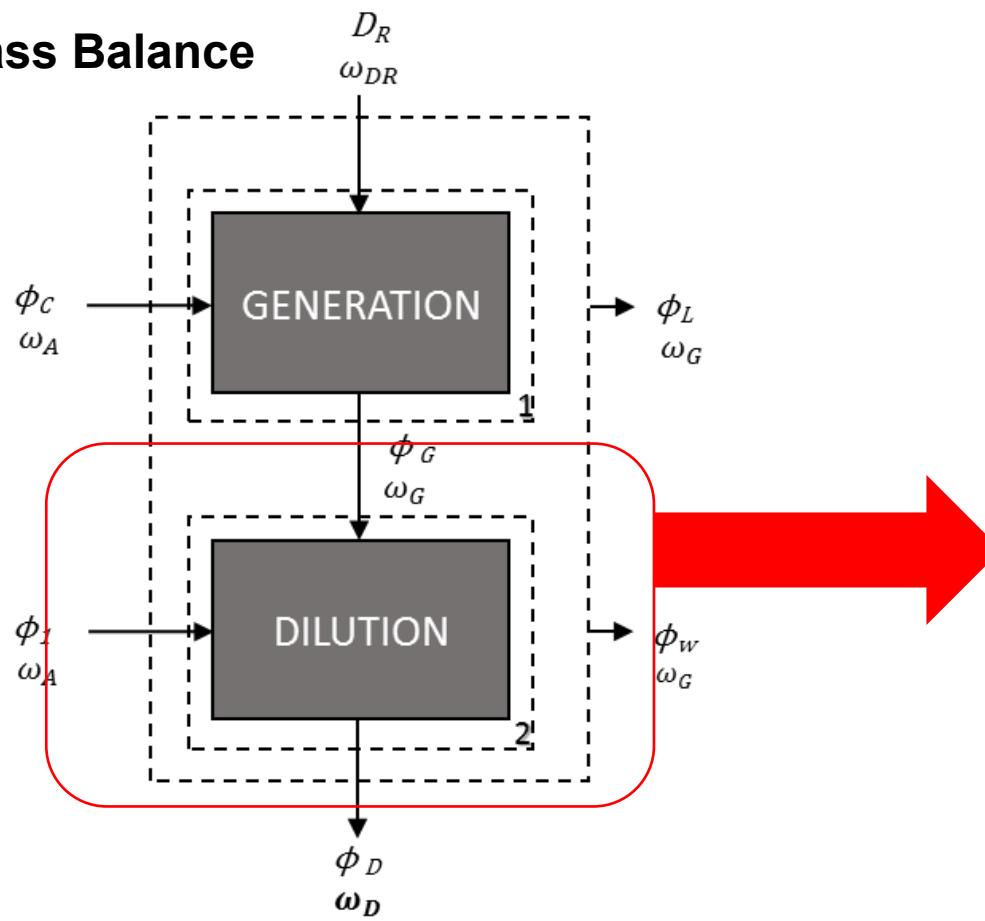
Mass Balance



Lecuna et al, 2018

Diffusion dynamic method

Mass Balance



Sassi et al, 2015

Diffusion based - Primary device

Gravimetric Method

Primary = Directly traceable to SI (kg)



✓ Average Diffusion rate

$$D_{R,average} = \omega_{VOC} \left(\frac{\Delta m + \Delta m_B}{\Delta t} \right)$$

Mass evaporated (corrected for buoyancy) over a period of time

✓ Actual Pressure Correction

$$D_{R,actual} = D_{R,average} \frac{\ln \left(1 - \frac{p_{VOC}}{p_{actual}} \right)}{\ln \left(1 - \frac{p_{VOC}}{p_{mean}} \right)}$$

Corrected to account for pressure variations

Sassi et al, 2011

Diffusion rate Uncertainty

Propagation of uncertainty GUM JCGM 100:2008 (E) by BIPM

X	[X]	x2	u(x)	u % (x)	SI [%]
Δm	g	0.01	3.81E-04	3.8%	100.0%
Δt	days	7	255	0.0%	0.0%
p_{mean}	Pa	98000	58	0.1%	0.0%
p_{actual}	Pa	101325	58	0.1%	0.0%
p_{VOC}	Pa	32154	643	3.0%	0.0%
ω_{VOC}	g.g ⁻¹	1.0	0.0	0.1%	0.0%
$d p_{actual}$	Pa	0	50	0.0%	0.0%
$d T_{actual}$	K	0	0.02	0.0%	0.0%
D_R	$\mu\text{g} \cdot \text{min}^{-1}$	0.96	0.036	3.8%	(k = 1)

$$u(\chi_{VOC}) = U(u(D_R), u(\phi_{dilution}))$$

Contribution to $u(x)$

$$Cx_i = \left(\frac{df}{dx_i} \right)^2 u^2(x_i)$$

Significance Index (SI)

$$SI_i = \frac{Cx_i}{Max Cx_i}$$

Diffusion based - Working standard

Portable generation

Working standard = Traceability to SI through a primary standard

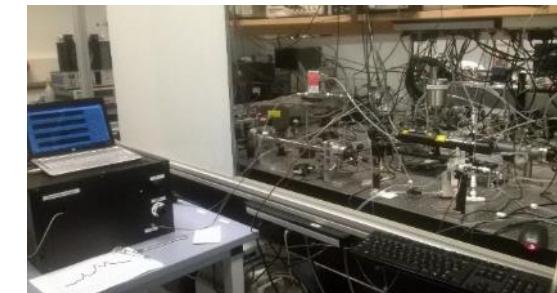


**PRIMARY or
TRANSFER STANDARD**

**WORKING
STANDARD**



Transportable device in INRIM (Italy)

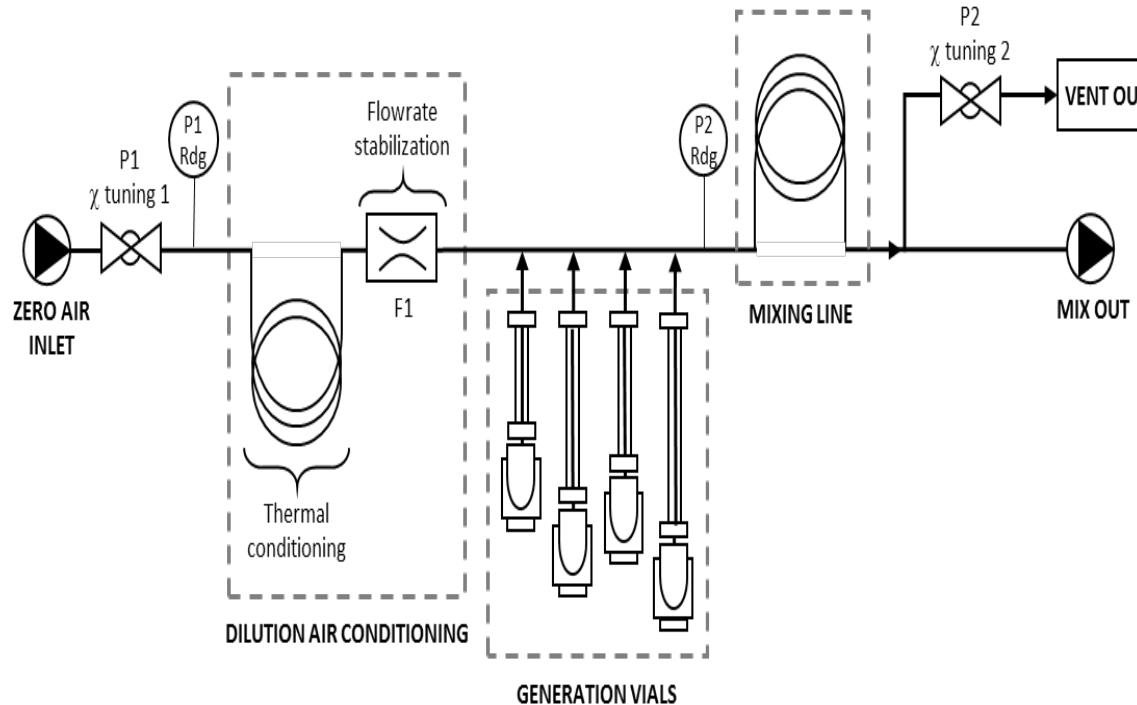


Transportable device in VSL (Netherlands), 2017

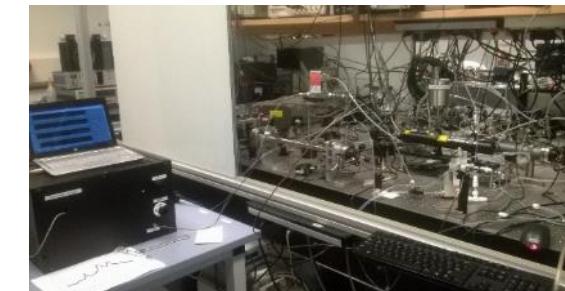
Diffusion based - Working standard

Portable generation

Working standard = Traceability to SI through a primary standard



Transportable device in INRIM (Italy)

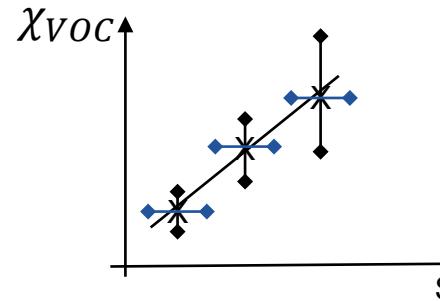


Transportable device in VSL (Netherlands), 2017

Demichelis et al, 2018

Working standard Calibration

Calibration curve of working standard, f



$$\chi_{VOC} = f(T, P, \phi_{dilution}, S)$$

**REFERENCE
GAS MIXTURE
(RGM)**



**UNKNOWN
AMOUNT
FRACTION
(WS)**

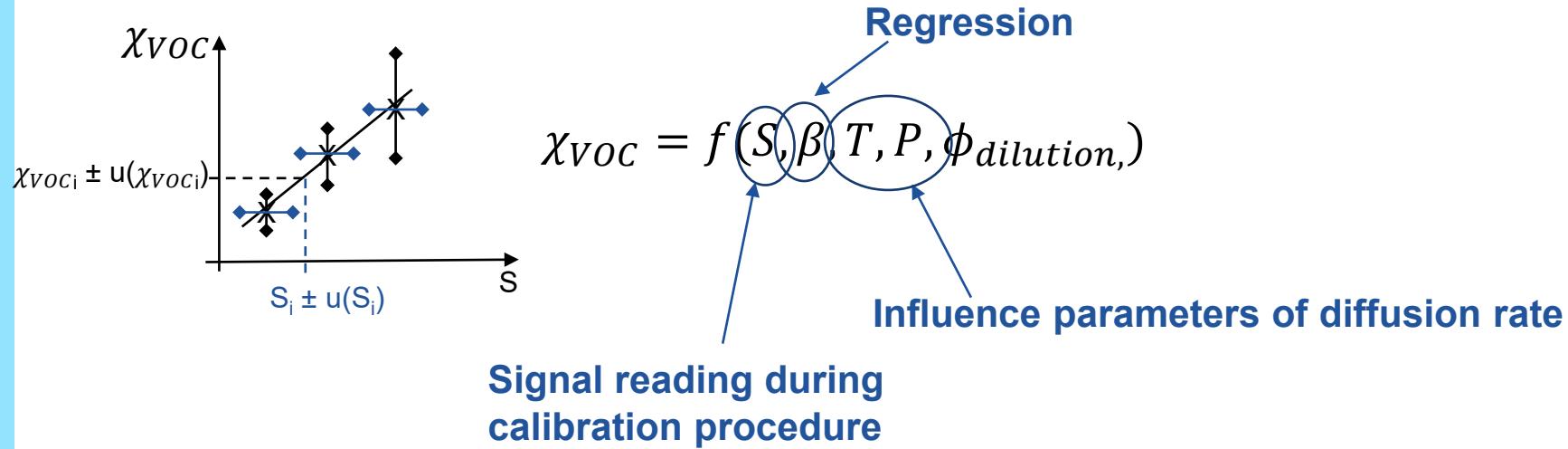


ANALYZER

**SIGNAL S_{RGM}
SIGNAL S_{ws}**

Working standard calibration

Calibration curve of working standard, f



Regression: Ordinary, Weighted, Weighted total depending on system

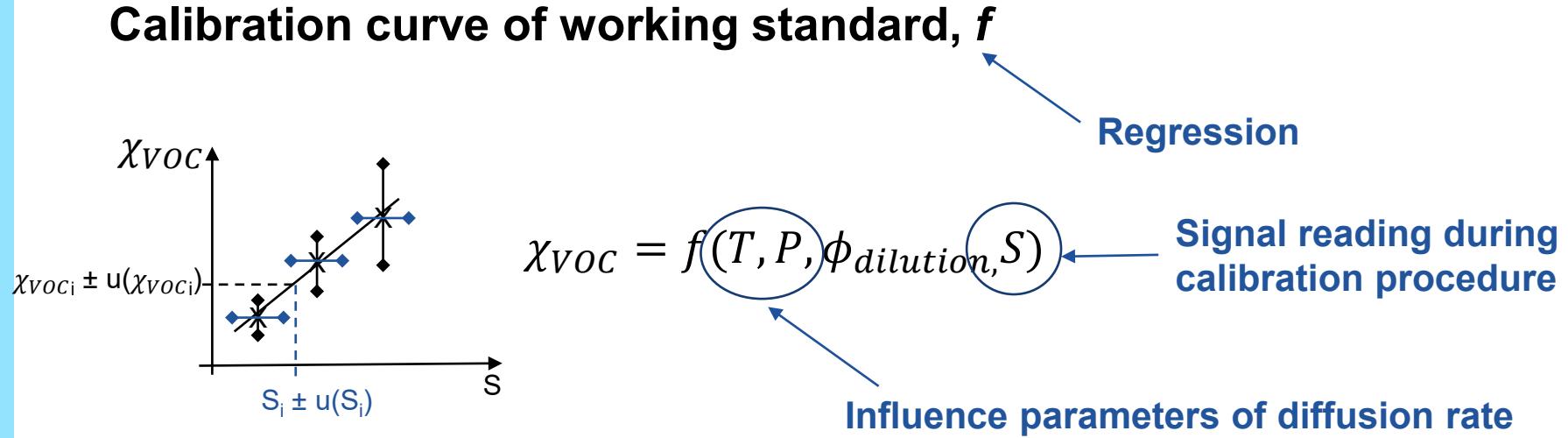
Uncertainty of regression: Algebraic or numerical (Montecarlo) approaches

NMI reference software for regression:

CCC Software (INRIM) (Lecuna et al 2020)

XGENLINE/ XLGENLINE (NPL)

Amount fraction Uncertainty



Uncertainty of the Calibration curve of working standard, f

$$U \chi_{VOC_{ws}}(T, P, \phi_{dilution}) = U(u(RGM), u(S), u(\beta_{regression}))$$



Metrology for Climate Relevant VOCs

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For more information, visit

[**www.metclimvoc.eu**](http://www.metclimvoc.eu)



The EMPIR initiative is co-funded by the European Union's Horizon 2020 research and innovation programme and the EMPIR Participating States